Name	Date Due	

Corrosion		
1. What is corrosion?		
Corrosion is a	where	
metals degrade by r	eacting	
chemicals i	n the	
environment such _	oxygen.	
2. What is rusting?		
Rusting is the name	given to	
corrosion of in	on.	
corrosion of in	on.	
3. Write down three ways that we corroding		
3. Write down three ways that w	ve can prevent metals	
3. Write down three ways that w	ve can prevent metals	
3. Write down three ways that we corroding  We can prevent me	ve can prevent metals	
3. Write down three ways that we corroding  We can prevent me	ve can prevent metals	
3. Write down three ways that we corroding  We can prevent me	ve can prevent metals tals	
3. Write down three ways that we corroding  We can prevent me	ve can prevent metals tals	

4. Sometimes metals are galvanised to prevent corrosion. What metal is used in this process?  The metal used in galvanisation is
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5. Cars rust (corrode) fast summer. Why is this?	ter in the winter than the
Cars	faster in the
winter than	summer as the
salt used to	the roads
speeds up the	process.