

Summary – “Chemicals”

Signs of a Chemical Reaction

The three **main signs** that a chemical reaction has happened are a **colour change**, **solid produced** (goes cloudy) or a **gas is given off** (fizzing/bubbles).

Chemical Reactions and Energy

A chemical reaction often turns **chemical energy** into **light**, **sound** and **heat** energy.

Elements

The **elements** are the **simplest “building blocks”** that make up us, the earth and the universe. They are listed in the **Periodic Table**.

Metals and Non-Metals

You can mark the **metals** and the **non metals** by drawing a **staircase** down the periodic table (see over). **Metals conduct** electricity. Most **non-metals do not conduct** electricity **apart from carbon** which does.

Groups (columns) of elements

The elements are arranged in “**families**” or “**groups**” which are the **columns** in the Periodic Table. They have numbers from 1-8/0. Some of the families also have names.

Alkali Metals

The **alkali metals** are **group 1** in the Periodic Table. The **react violently with water** producing **hydrogen gas** and the **hydroxide** of the alkali metal.

e.g. lithium + water → lithium hydroxide

Noble Gases

The **noble gases** are group **8/0** of the Periodic Table. Most of the time, they **do not react with anything** which is why they are called “noble”. They are often used in lights

Halogens

The **halogens** are **group 7** of the Periodic Table. They are quite reactive.

Compounds and Mixtures

A **compound** is **more than one element joined together**. In a **mixture**, the **elements are not joined together**. Compounds with **only two elements** in them get their names from the **name of the two elements with “ide” at the end**. If there is a **metal**, it is **written first**. If a compound name ends in “**ate**”, then it contains **oxygen**.

e.g. iron + oxygen → iron oxide

e.g. sulphur + chlorine → sulphur chloride

e.g. iron + carbon + oxygen → iron carbonate

Writing down Chemical Reactions

A chemical reaction can be **written** with the **reactants** (things you start with) on the **left** and the **products** (things you end up with) on the **right** with an **arrow** pointing between them.

e.g. hydrogen + oxygen → hydrogen oxide

The Periodic Table

Metals → Non-metals

Lanthanum	Cerium	Praseodymium	Neodymium	Promethium	Samarium	Europium	Gadolinium	Dysprosium	Terbium	Holmium	Erbium	Thulium	Ytterbium	Yb
La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Ytterbium	
57	58	59	60	61	62	63	64	65	66	67	68	69	70	
Lanthanum	Cerium	Praseodymium	Neodymium	Promethium	Samarium	Europium	Gadolinium	Dysprosium	Terbium	Holmium	Erbium	Thulium	Ytterbium	
89	90	91	92	93	94	95	96	97	98	99	100	101	102	No
Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Md	Fm	Mendelevium	Nobelium
Actinium	Thorium	Protactinium	Uranium	Neptunium	Plutonium	Americium	Curium	Berkelium	Californium	Einsteinium	Fermium	Mendelevium	Nobelium	