Nature's Chemistry 1

Fuels - The "Burning" Question



Name			
Class	 	 	
Teacher			

Aim: What is a Fuel?

"Most fuels release energy after they ignite and react with oxygen in the air (often known as "burning" or "combustion". This energy that is stored in fuels can then be used to make things move such as vehicles used for transport or generators to make electricity. There are various other exothermic chemical reactions and nuclear reactions, such as nuclear fission or nuclear fusion that use fuels to produce energy. Fuels are also used in the cells of organisms in a process known as "respiration", where organic molecules react to release usable energy. The most common types of chemical used as fuels are "hydrocarbons", but many other substances, such as radioactive metals, are currently used as well."

1. What is a fuel?		
A fuel is a substance produce energy. This to make	_ move.	_ burns to can be used

2. How is the energy in fuels released?		
The energy in fuels is	by burning	
or combusting them.	involves	
igniting them and allowing	to	
react with oxygen in	air.	

3. What is the energy in fuels used for?	
The energy in fuels can use	d to make
things move as cars, buses,	planes or
trains can also be used to	
electricity by using a gene	rator.

Aim: What do we use fuels for?

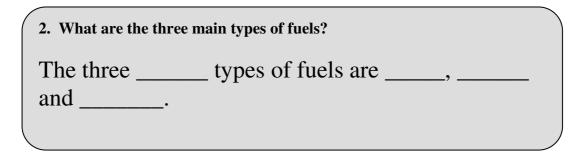
The first use of fuel was burning wood or sticks for heat and cooking by our ancestors nearly 2 million years ago. Throughout the majority of human history fuels made from plant or animal fat were the only ones available for human use. Charcoal, which comes from wood, has been used since at least 6,000 BCE to extract metals from their ores. It was only replaced by coke, derived from coal, as the forests started to become depleted around the 18th century. Charcoal briquettes are now commonly used as a fuel for barbecue cooking.

Coal was first used as a fuel around 1000 BCE in China. With the development of the steam engine in 1769, coal came into more common use for driving ships and trains. By the 19 th century, gas extracted from coal was being used for street lighting in London. In the 20th century, the primary use of coal was for the generation of electricity, providing 40% of the world's electrical power supply in 2005.

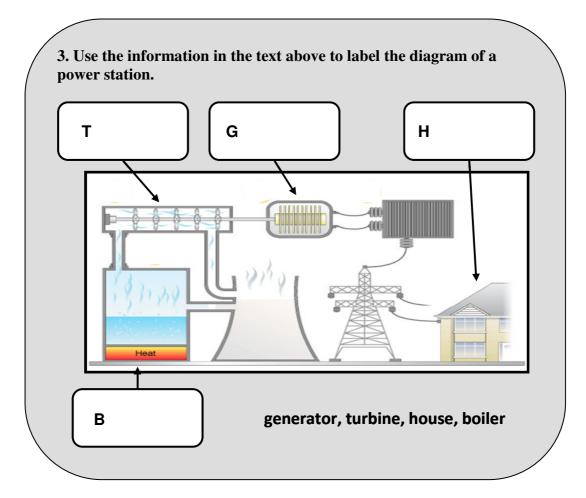
1. Give	e three uses of fuels	
Fuels	can be used for, and,	
2. Uso	e the information in the video to answer these questions? What fuel does the power station in the video use?	
station	What is the name of the Canadian province where the power n is?	
	What do the pulverisers do?	
	What does the steam do to the turbines?	
	What is the turbine attached to?	
	What particles flow to create electricity?	

Aims: What types of fuels are there? How do we make electricity?

c. What combine to make a railroad? and S E d. What was burned to make electricity? e. What fuel did the Wright Brothers use for Aviation? f. What was the first fossil-fuelled conflict? VW g. What happened in the 70s? S h. Who is burning half the world's coal? i. What does the video say we need to learn to live without F j. What is the third fuel type mentioned?	c. What combine to make a railroad? and SE d. What was burned to make electricity? e. What fuel did the Wright Brothers use for Aviation? f. What was the first fossil-fuelled conflict? VW g. What happened in the 70s? S h. Who is burning half the world's coal? i. What does the video say we need to learn to live without? F j. What is the third fuel type mentioned?		What fuel did we burn instead?
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j. What is the third fuel type mentioned?	j. What is the third fuel type mentioned? G	i.	What does the video say we need to learn to live without?
G		j.	
			In what way do you agree/disagree with the video?



Fuels such as coal, oil and gas are burnt to produce heat energy. This makes water boil to produce steam which drives turbine, making it rotate. The rotating turbine turns a generator which produces electricity for us to use in our homes.

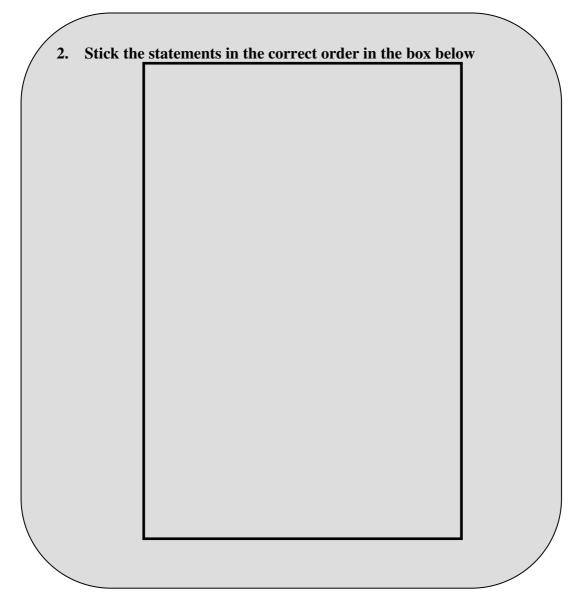


4 Complete the table with the parts of the power station

Part	What it does	What would happen if it was missing?
Boiler		
Turbine		
Generator		

Aim: How are fossil fuels formed?

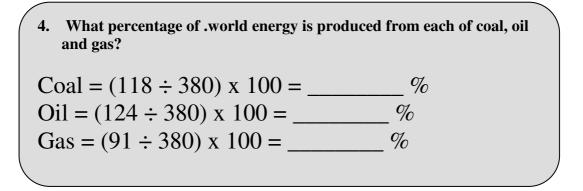
1. Fossil Fuel Formation	
a. How many people are on plan	et earth?
billion	
b. Where does most energy begin	n its journey?
The S	
c. Where do we need to dig?	
D into the u 1	of our p
d. What is the first thing to reali	
Not all a a	and
p become f	f
e. What is an example of a satur	
A s	
f. What does "anaerobic" mean	
No o aroun	d
g. What is pressure?	
A f pushing d	on or
a something	
h. What are the 4 things needed fossil fuel?	for something to become a
A s environ	nent, a
conditions, lots of p	and high t
i. What areas have the ideal con	
Γareas	
j. What did the changes in the e	nvironment cause
S l to r	
k. What happened without oxyg	
The p did not full	y r a
l. What were the layers a bit lik	e?
A s	
m. Where do we see layers of coa	
U in parts of S	



Something that is *infinite* will never end or run out. Something that is *finite* will end or run out at some point

3. Are fossil fuels finite or infinite? Explain your answer.

In 2012, the total energy produced was 380 BTU (British Thermal Units). 118 BTU was from coal, 124 BTU from oil and 91 BTU from gas.



Aim: What is made when fuels are burnt?

Air is made of 78 % nitrogen gas, 20 % hydrogen gas and 2 % carbon dioxide gas. When things burn, they are reacting with oxygen in the air.

1. What are the names of the three gases in the air? The gases in the air oxygen, nitrogen and c dioxide.	
2. Which gas in the air is used when fuels burn? When fuels, oxygen is used.	

When most fuels are burnt, carbon dioxide and water are produced. A chemistry would write this in the **word equation**:

 $Fuel + oxygen \rightarrow carbon \ dioxide + water$

If the fuel is "methane" (natural gas), this can be written as:

Methane + $oxygen \rightarrow carbon \ dioxide$ + water

3. Write a word equation for the burning of each of the fuels "ethane", "butane", "propane" and "dodecane".
_______ + oxygen → carbon dioxide + water
Butane + oxygen → ______ + water
Propane + _____ → carbon dioxide + water
Dodecane + oxygen → carbon dioxide + _____

Instead of writing words, we can use **chemical formulae** to write the equation. The formula for methane is " CH_4 ", oxygen is " O_2 ", carbon dioxide is " CO_2 " and water is " H_2O ". This leads to the **formula equation**:

 $CH_4 + O_2 \rightarrow CO_2 + H_2O$