1. **copy and complete the following examples**

 Name of compound formula

1. nitrogen monoxide
2. silicon tetrachloride
3. boron trifluoride
4. tetraarsenic hexaoxide
5. disulphur dichloride
6. **write out the chemical formulae for the following 2 element compounds**
	1. Boron Fluoride
	2. Calcium iodide
	3. Sodium sulfide
	4. Lithium nitride
	5. Tin bromide
	6. Strontium bromide
7. Write simple chemical formulae for the following
8. Potassium carbonate
9. Lithium hydrogencarbonate
10. Sodium dichromate
11. Ammonium bromide
12. Lithium hydrogensulphite
13. Potassium sulfate
14. Ammonium nitrite
15. Sodium carbonate.
16. Write simple chemical formulae for the following
17. Nickel (II) sulfide
18. Tin (IV) chloride
19. Lead (II) iodide
20. Silver (II) fluoride
21. Copper (I) oxide
22. Uranium (VI) fluoride
23. Manganese (VII) oxide
24. Bismuth (III) hydride
25. Write simple chemical formulae for the following
26. Calcium hydrogencarbonate
27. Ammonium phosphate
28. Magnesium nitrate
29. Calcium nitrite
30. Ammonium sulfite
31. Barium hydrogensulfate
32. Aluminium hydroxide
33. Aluminium sulfate